



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>5</sup> :B41M 5/40, 5/00, 5/38  
G03F 7/20

A1

(11) International Publication Number:

WO 93/25392

(43) International Publication Date: 23 December 1993 (23.12.93)

(21) International Application Number: PCT GB93 01267

(22) International Filing Date: 15 June 1993 (15.06.93)

(74) Agents: MILLROSS, Christopher, Robert et al.; ICI  
Group Patents Services Dept., P.O. Box 6, Shire Park,  
Welwyn Garden City, Hertfordshire AL7 1HD (GB).

(30) Priority data:

9212630.9

15 June 1992 (15.06.92)

GB

9216277.5

30 July 1992 (30.07.92)

GB

(81) Designated States: JP, US, European patent (AT, BE, CH,  
DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT,  
SE).(71) Applicant (for all designated States except US): IMPERIAL  
CHEMICAL INDUSTRIES PLC [GB, GB]; Imperial  
Chemical House, Millbank, London SW1P 3JF (GB).

Published

With international search report.

(72) Inventors; and

(75) Inventors/Applicants (for US only): HUTT, Kenneth, West  
[GB, GB]; Dysert, Colchester Road, Wix, Nr Manning-  
tree, Essex CO11 2RT (GB). STEPHENSON, Ian, Ri-  
chard [GB, GB]; 10 Pardoe Place, Rushmere, St. An-  
drews, Ipswich IP4 5UH (GB).

(54) Title: THERMAL TRANSFER PRINTING RECEIVER SHEET AND METHOD

(57) Abstract

A method of light induced thermal transfer printing which comprises printing directly onto a substantially apolar material for example a polyolefin without the need for a release agent in the dye-receptive surface and imaged receiver sheets comprising a dye receptive surface containing a substantially apolar material.

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	FR	France	MR	Mauritania
AU	Australia	GA	Gabon	MW	Malawi
BB	Barbados	GB	United Kingdom	NL	Netherlands
BE	Belgium	GN	Guinea	NO	Norway
BF	Burkina Faso	GR	Greece	NZ	New Zealand
BG	Bulgaria	HU	Hungary	PL	Poland
BJ	Benin	IE	Ireland	PT	Portugal
BR	Brazil	IT	Italy	RO	Romania
CA	Canada	JP	Japan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SK	Slovak Republic
CI	Côte d'Ivoire	LJ	Liechtenstein	SN	Senegal
CM	Cameroon	LK	Sri Lanka	SU	Soviet Union
CS	Czechoslovakia	LU	Luxembourg	TD	Chad
CZ	Czech Republic	MC	Monaco	TG	Togo
DE	Germany	MG	Madagascar	UA	Ukraine
DK	Denmark	ML	Mali	US	United States of America
ES	Spain	MN	Mongolia	VN	Viet Nam
FI	Finland				

Thermal Transfer Printing Receiver Sheet and Method

This invention relates to a thermal transfer printing method and to an imaged receiver sheet produced by the method.

Thermal transfer printing is a printing process in which a dye is caused, by thermal stimuli, to transfer from a dyesheet to a receiver sheet. In such processes, the dyesheet and receiver sheet are placed in intimate contact, the thermal stimuli are applied to the dyesheet and the dyesheet and receiver sheet are then separated. By applying the thermal stimuli to pre-determined areas in the dye-sheet, the dye is selectively transferred to the receiver to form the desired image.

The thermal stimuli were originally provided by means of mechanical print heads incorporating small electrical heaters. More recently, however, the thermal stimuli have been provided by heat inducing light, for example a laser.

Receiver sheets conventionally comprise a substrate with a dye-receiving polar surface on one side, into which a dye is thermally transferable and retainable. Where the substrate is itself polar and capable of receiving a dye, the dye may be transferred directly to a surface of the substrate. However receiver sheets typically comprise a substrate supporting a receiver layer specifically tailored to receive the dye. Receiver sheets may also comprise a backcoat to impart desirable characteristics including improved handling properties and to provide for improved ease of lamination. It is known to use polar polymers, for example, polyester and polyvinyl alcohol/polyvinylchloride copolymer, in the receiver layer as such polymers provide for good dye compatibility.

By "polar polymers" we mean a polymer comprised of monomeric units which have a significant dipole moment.

Dye sheets conventionally comprise a substrate having on one side thereof, a dye layer comprising a thermally transferable dye dissolved and/or dispersed in a polymeric binder, which again is a polar polymer. Where heat inducing light is to be employed, it is necessary to employ a light-absorbing material which may be present in the dye coat or may be present in a separate layer preferably located between the dye coat and the substrate in order to allow absorption of the inducing light.

However, as the dyesheet and receiver sheet are in intimate contact during the printing process and high temperatures are employed, problems with

SUBSTITUTE SHEET

the polar substrate or receiver layer and the dyesheet bonding together. at least temporarily, have been encountered. A consequence of this bonding is that further undesirable transfer of dye from the dyesheet to the receiver sheet may occur in addition to the desired dye transfer causing loss of tone control and, in extreme cases where other components of the dye coat transfer, production of a half tone rather than a continuous tone image.

This problem has been addressed by employing a further component in the receiver layer to act as a release agent to facilitate the separation of the polar surface of the receiver sheet and the dye sheet at the required time, thus reducing the risk of unwanted dye transfer. Examples of release agents include silicone polymers for example siloxanes.

The provision of a receiver layer on the substrate allows a release agent to be incorporated in the dye-receptive surface but may lead to an increase in the cost of production of receiver sheets. Further, the presence of a release agent as an extra component in a receiver layer may also complicate the formulation and coating of the receiver layer onto the substrate.

An increasing use of thermal transfer printing is in the preparation of prints from images generated by a colour video camera. Clearly, it is important that the quality of such prints be equal to that of prints provided by conventional silver halide photographic methods. In particular, a reflective optical density of at least 1.6 is required for commercial acceptance. In addition, it is advantageous if a thermal transfer print has the same feel as a photographic print.

It is well known that photographic prints are produced on paper having a surface coating of a polyolefin. Such paper is relatively cheap, is readily available and would, of course give the same feel as photographic prints. Hence, there would be considerable advantages in using it for the preparation of thermal transfer prints. In particular, as polyolefins are apolar polymers( ie polymers comprised of monomeric unit which do not have a significant dipole moment) any tendency to bond to the polar material of the dye sheet is likely to be less.

GB Patent Specification No 2 217 866 does, in fact, disclose the use of such paper for thermal transfer prints using a heated print head as the source of the thermal stimuli. However, there is no disclosure of the optical density attained and tests have shown that the maximum O.D.

achievable is only circa 50% of the commercially acceptable value. Moreover the prints have a tendency to fade with time. Hence the use of such paper with "conventional" thermal transfer printing is not viable.

As mentioned previously, lasers are now being used as the source for the thermal stimuli and it is known that thermal printing using a laser produces an increase in O.D. of only circa 30% when used with a conventional receiver sheet. Hence, it would be expected that the best O.D. obtainable with a laser and an apolar receiver sheet would be of the order of 1.4, still substantially below the commercially viable value.

Surprisingly, we have now found that prints with an O.D. in excess of 1.6 are obtainable using a apolar receiver sheet when the thermal stimuli are produced by heat inducing light.

According to a one aspect of the invention there is provided a method of light-induced thermal transfer printing which comprises providing a receiver sheet having a dye-receptive surface and a dye sheet having a light absorbing material for converting the inducing light into thermal energy and a dye coat comprising a binder and a dye, arranging the dyesheet on the receiver sheet such that dye coat and the said surface are adjacent, preferably in intimate contact, applying inducing light to the dyesheet which is absorbable by the light absorbing material to produce thermal stimuli in pre-determined areas of the dye sheet thereby to cause dye to transfer from the dye coat to the surface and separating the dyesheet and the receiver sheet wherein the said surface contains, as a dye-receptive medium, a substantially apolar organic polymeric material.

By "dye-receptive surface" we mean the volume of the receiver sheet defined by at least part of the surface thereof and the depth of the sheet at that surface to which a dye may be transferred in a thermal transfer printing process.

Suitably, the substantially apolar material comprises an organic polymer having monomeric units a significant proportion of which do not comprise halogens and/or functional moieties comprising heteroatoms.

The invention further provides an imaged receiver sheet comprising a substrate optionally supporting a receiver layer which substrate (or, if present, receiver layer) has a dye-receptive surface which contains, as a dye-receptive medium, a substantially apolar polymeric organic material containing a thermally transferable dye defining an image wherein the said

dye is transferred to said dye-receptive surface by means of a light induced thermal transfer printing process.

A method and imaged receiver sheet according to the invention is advantageous as it enables a dye to be retained in the substantially apolar material of the receiver sheet without the need for the presence of a polar material to provide the appropriate dye compatibility and a dye image of acceptable quality may be secured. Further, the substantially apolar polymeric material in the receiver sheet has less of a tendency to stick to the dye sheet with which it is in contact during printing hence there is no need for a release agent as may be required with conventional receiver sheets to facilitate the release of the two sheets. A further benefit of the invention, as a consequence of not requiring a release agent, is that it is not necessary to provide a separate receiver layer (in which the release agent is typically located) on the substrate. An additional important advantage is that the feel of the final print will be substantially the same as a photographic print.

However, we do not exclude the case where a receiver layer is present on the substrate of a receiver sheet of the invention provided that the said layer contains, as a dye-receptive medium, a substantially apolar polymeric organic material. Thus, the presence of a receiver layer is optional rather than being necessary to accommodate a release agent. Furthermore, the nature of the receiver layer need not be constrained by the requirement that it be compatible with a release agent.

The substantially apolar material contained in the dye-receptive surface suitably comprises a polyolefin, a mixture and/or a block or random copolymer of polyolefins. Preferred polyolefins include polypropylene and polyethylene, isotactic polypropylene being particularly preferred. A blend of polyolefins may be employed, and the relative amounts of the components of the blend may be selected to provide desirable porosity and surface gloss characteristics.

The substantially apolar material may be solid, porous or voided as desired.

Suitably the substantially apolar polymeric organic material is present in the dye-receptive surface in an amount by weight greater than any other organic component in the surface, preferably in an amount of at least 50%.

especially at least 80% and desirably substantially 100% by weight of organic components in the said surface.

The dye-receptive surface preferably also comprises an inorganic filler to impart opacity which is especially desirable if the receiver sheet comprises a substrate without any receiving layer and to provide improved wear characteristics and optical density. Suitable fillers include alumina, silica and particularly titanium dioxide.

A particularly preferred receiver sheet according to the invention comprises an isotactic polypropylene substrate having a filler comprising titanium dioxide.

The inorganic filler may be present in an amount of 0.5 to 50 % and preferably 1 to 20% by weight of the dye-receptive surface.

The dye-receptive surface may comprise other organic materials, as a minor proportion of the total organic materials in the dye-receptive surface, in addition to the substantially apolar material in order to impart desired characteristics to the surface. Suitable organic materials include polymers for example polyvinyl chloride, polyacrylonitrile, polystyrenes and polyesters, and also include the monomeric compounds of suitable polymers including acrylonitrilebutylstyrene (ABS). Suitably the apolar material is present in the dye receptive surface in an amount of at least 50% and preferably at least 80% by weight of the total organic components in the said surface.

Where the receiver sheet comprises a substrate having a dye-receptive surface without a separate receiver layer, the substrate is desirably adapted by the provision of a smooth surface texture.

However, the substrate may have a receiver layer on one side of the substrate, which layer comprises a dye-receptive surface comprising a substantially apolar polymeric organic material into which thermally transferable dyes can readily pass in a TTP process.

It is not necessary for the receiver layer to contain a material which acts as a release agent, preferably the receiver layer is substantially free of a release agent.

Where the receiver sheet comprises a substrate and a receiver layer having a dye-receptive surface, substrates known in the art may be employed in the present invention including cellulose fibre paper, synthetic paper for example TYVEK synthetic paper, thermoplastic films for example polyethylene

terephthalate (desirably biaxially orientated), filled and/or voided thermoplastic films for example pearl film, and laminates of two or more substrate materials.

5 It is essential that the dye-receptive surface of the receiver layer comprises a substantially apolar organic polymeric material. However, that part of the receiver layer which does not function as the dye-receptive surface may comprise a conventional receiver layer material for example polar polymers as described herein.

10 The substrate may also have a back coat on the opposite side to the dye-receiving surface, if desired, to impart desirable properties for example, to improve handling characteristics and to aid adhesion of a protective cover sheet to the receiver sheet.

15 Suitably, the back coat, if present, comprises a cross-linked polymer binder and is provided to impart desirable properties to the receiver sheet for example improved handling characteristics and reduced tendency to retransfer the dye at low temperatures. If desired, the back coat may have a textured surface which may be imparted by a filler material or by the polymer per se.

20 Suitably a receiver sheet made according to the present invention may be laminated with a cover sheet on both sides to provide protection for the image on the sheet so that the receiver sheet is suitable for use as for example as a security or identity badge. The cover sheet may be the same or different on the different sides of the sheet and is preferably transparent on at least one side of the sheet. The cover sheet suitably comprises a 25 thermoplastic film, for example polyvinyl chloride, polyethylene terephthalate and polycarbonate compositions. When used in such a way, the receiver sheet may be in the form of a film of apolar polymeric material.

The inducing light is desirably a laser, for example Nd:YAG, Argon ion and Ti:sapphire and preferably a laser diode.

30 For lasers operating in the near infra-red, there are a number of organic materials known to absorb at the laser wavelengths. Examples of such materials include the substituted phthalocyanines described in EP-B-157,568, which can readily be selected to match laser diode radiation at 750-900 nm, for example and carbon black pigment which has a broad absorption spectrum 35 and is thus useful for a wide range of visible light and infra red emitting lasers.



Also of importance is the provision of sufficient absorber for the system used. It is desirable to use sufficient to absorb at least 50% of the incident inducing light. We prefer to use sufficient to absorb at least 90% of the inducing light, to obtain an optical density of 1 in transmission, although higher proportions may be used if desired.

A variety of materials can be used for the dyesheet substrate, including transparent polymer films of polyesters, polyamides, polyimides, polycarbonates, polysulphones, polypropylene and cellophane, for example. Biaxially orientated polyester film is the most preferred, in view of its mechanical strength, dimensional stability and heat resistance. The thickness of the substrate is suitably 1-50  $\mu\text{m}$ , and preferably 2-30  $\mu\text{m}$ .

Any dye capable of being thermally transferred may be selected as required. Dyes known to thermally transfer, come from a variety of dye classes, eg from such nonionic dyes as azo dyes, anthraquinone dyes, azomethine dyes, methine dyes, indoaniline dyes, naphthoquinone dyes, quinophthalone dyes and nitro dyes. The dyecoat binder can be selected from such known polymers as polycarbonate, polyvinylbutyral, and cellulose polymers, such as methyl cellulose, ethyl cellulose and ethyl hydroxyethyl cellulose, for example, and mixtures of these.

The dyecoat may also include dispersing agents, antistatic agents, antifoaming agents, and oxidation inhibitors, and can be coated onto the absorber layer as described for the formation of the latter. The thickness of the dyecoat is suitably 0.1-5  $\mu\text{m}$ , preferably 0.5-3  $\mu\text{m}$ .

The dyesheet may be elongated in the form of a ribbon and housed in a cassette for convenience, enabling it to be wound on to expose fresh areas of the dyecoat after each print has been made.

Dyesheets designed for producing multicolour prints have a plurality of panels of different uniform colours, usually three: yellow, magenta and cyan, although the provision of a fourth panel containing a black dye, has also previously been suggested. When supported on a substrate elongated in the form of a ribbon, these different panels are suitably in the form of transverse panels, each the size of the desired print, and arranged in a repeated sequence of the colours employed. During thermal transfer printing, panels of each colour in turn are held against a dye-receptive surface of the receiver sheet, as the two sheets are imagewise selectively heated the first

colour being overprinted by each subsequent colour in turn to make up the full colour image.

In applying a coating to a substrate, either for the dye sheet or the receiver sheet, various coating methods may be employed including, for example, roll coating, gravure coating, screen coating and fountain coating. After removal of any solvent, the coating can be cured for example by heating or by irradiation with for example ultra violet light, electron beams and gamma rays.

The invention will now be illustrated by way of the following non-limiting examples.

#### Example 1

A receiver sheet according to the invention was prepared by coating isotactic polypropylene onto a paper substrate to a coat weight of 15gsm.

A selection of dye sheets were prepared as follows:

#### Dyesheets 1, 2, 3

6 $\mu$ m polyester film (Toray) having a back coat and a sub-coat was coated to a dry coat thickness of approximately 1 $\mu$ m by gravure printing (yellow and magenta) and 2.5 $\mu$ m using a Meyer bar (cyan) with the following dye compositions (amounts are in kg unless otherwise stated) and the compositions were dried by heating in air for about 15 seconds at 110°C:

		Dyesheet 1	Dyesheet 2	Dyesheet 3
		<u>Yellow</u>	<u>Magenta</u>	<u>Cyan</u>
25	thermal transfer dye (TTD) Y1	2.231	-	-
	TTD Y2	0.954	-	-
	TTD M1	-	2.505	-
	TTD M2	-	1.659	-
	TTD C1	-	-	0.865
	TTD C2	-	-	1.298
30	Binder: PVB-BX1	2.547	2.632	-
	EC-T10	0.638	0.657	0.541
	EC-T200	-	-	1.622
	Infra-red absorber	0.841	0.954	0.571
Tetrahydrofuran		50 litres	50 litres	50 litres

Y1 was C1 solvent yellow 141; Y2 was C1 disperse yellow 126; M1 was 3-methyl-4-(3-methyl-4-cyanoisothiazol-5-ylazo)-N-ethyl-N-acetoxyethyl aniline; M2 was C1 disperse red 60; C1 was 3-acetylamino-4-(3-cyano-5-phenylazothiophenyl-2-ylazo)-N,N-diethyl aniline; C2 was C1 solvent blue 63; PVB-BX1 was polyvinyl butyral BX1 from Sekisui; EC-T10 and EC-T200 were ethyl cellulose grade T10 and T200 respectively from Hercules.

The prepared dyesheets were each brought into contact with a sample of the receiver sheet according to the invention by application of 1 atmosphere pressure. An STC LT-100 laser diode operating at 807nm was collimated and then focused using a 160mm achromat lens. The incident laser power at the dyesheet was about 60mW and the laser spot (full width at half maximum power) was about 30x20 $\mu$ m. The laser spot was scanned by a galvanometer scanner. The dyesheet and receiver sheet were held on an arc which allowed focus to be retained throughout the scan length. The scanning equipment addressed the laser to locations 20x10 $\mu$ m apart giving a good overlap of adjoining pixels. At each pixel the laser was pulsed for a specific time of between 100 and 600 $\mu$ s according to the desired optical density of the pixel to provide high resolution dye pixels on the receiver sheets.

The optical density of the transmitted dye was recorded using a Sakura densitometer operating in reflection mode and the results are listed in Table 1.

#### Example 2

The test carried out in Example 1 was repeated using the same dye-sheets and printing process with a receiver sheet comprising isotactic polypropylene filled with 3% by weight of the receiver sheet of titanium dioxide (rutile form).

The optical density of the transmitted dye was recorded using a Sakura densitometer operating in reflection mode and the results are listed in Table 1.

#### Example 3

The test carried out in Example 1 was repeated using the same dye-sheets and printing process and using a polyethylene based synthetic paper (TYVEK) as a receiver sheet.

The optical density of the transmitted dye was recorded using a Sakura densitometer operating in reflection mode and the results are listed in Table 1.

Table 1

<u>Pulse Duration</u>	<u>Example 1</u>			<u>Example 2</u>			<u>Example 3</u>		
<u>μs</u>	<u>Yellow</u>	<u>Magenta</u>	<u>Cyan</u>	<u>Yellow</u>	<u>Magenta</u>	<u>Cyan</u>	<u>Yellow</u>	<u>Magenta</u>	<u>Cyan</u>
50	0.06	0.04	0.08	0.15	0.09	0.09	0.02	0.1	0.01
100	0.18	0.18	0.22	0.22	0.15	0.13	0.09	0.13	0.00
150	0.36	0.36	0.45	0.33	0.30	0.28	0.38	0.57	0.03
200	0.57	0.57	0.73	0.52	0.50	0.50	0.71	0.85	0.13
250	0.84	0.84	1.05	0.70	0.69	0.97	0.91	0.93	0.26
300	1.16	1.15	1.35	0.90	0.88	1.24	0.99	1.01	0.53
350	1.49	1.48	1.61	1.07	1.10	1.44	1.04	1.12	0.82
400	1.74	1.71	1.76	1.29	1.32	1.63	1.10	1.22	1.08
450	1.92	1.84	1.81	1.51	1.53	1.83	1.26	1.23	1.18
500	2.08	1.93	-	1.71	1.70	1.88	1.33	1.36	1.28
550	-	2.0	-	1.79	1.74	-	1.31	1.32	1.29

The above results demonstrate that suitable optical density may be obtained. Further, there was no evidence of total dye transfer in the above experiments despite the absence of a release agent in the dye-receptive surface of the receiver sheet and in the dye coat of the dye sheet.

Example 4

To indicate the advantages of the invention, Example 1 was repeated except that the printing of the magenta dye was carried using a programmable print head supplying heat pulses of 2 to 10 ms (ie 2000 to 10,000 µs). The results are shown in Figure 1 and the results for magenta from Example 1 are shown in Figure 2. An increase in maximum O.D. of from 0.8 to 2 (ie an increase of 250%) is apparent, whereas Figures 3 and 4, which show the measured O.D. for magenta using a conventional receiver sheet printed by in both ways, shows an increase in maximum O.D. of from 2 to 3 (ie an increase of only 50%).

Example 5

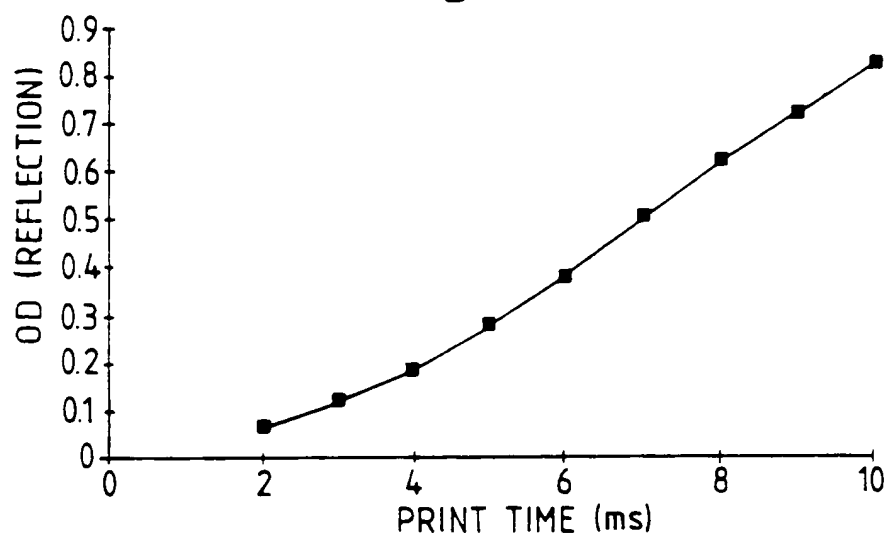
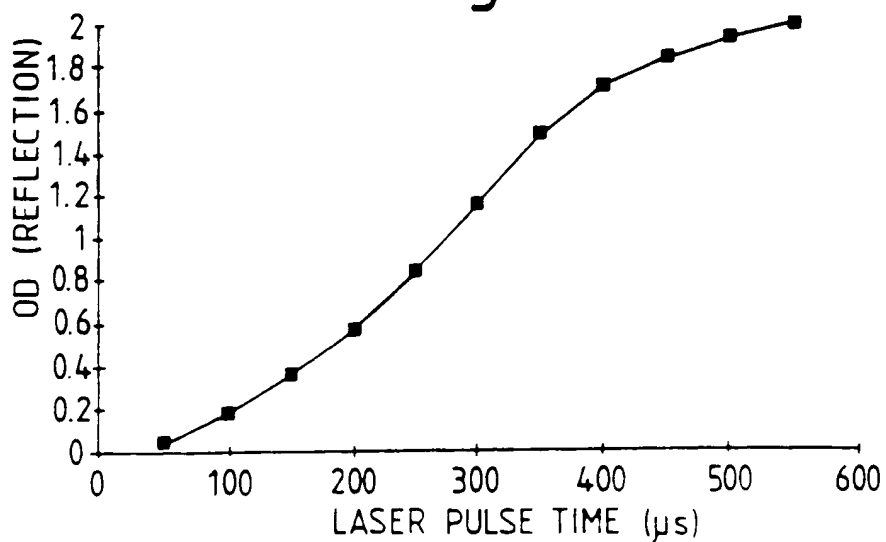
- 11 -

Ageing characteristics were investigated by storing the prints under ambient conditions for circa one year and then remeasuring the O.D. Figure 5 shows the results for prints made by the conventional print head method and Figure 6 shows the results for prints made in accordance with the invention. In the case of conventional printing, there is an appreciable drop in O.D. at original levels as low as 0.3 whereas with printing in accordance with the invention there is no drop until an O.D. of 1.2. Moreover, conventional prints suffer a drop in maximum O.D. of 30% whereas prints according to the invention only drop by 10%.

•

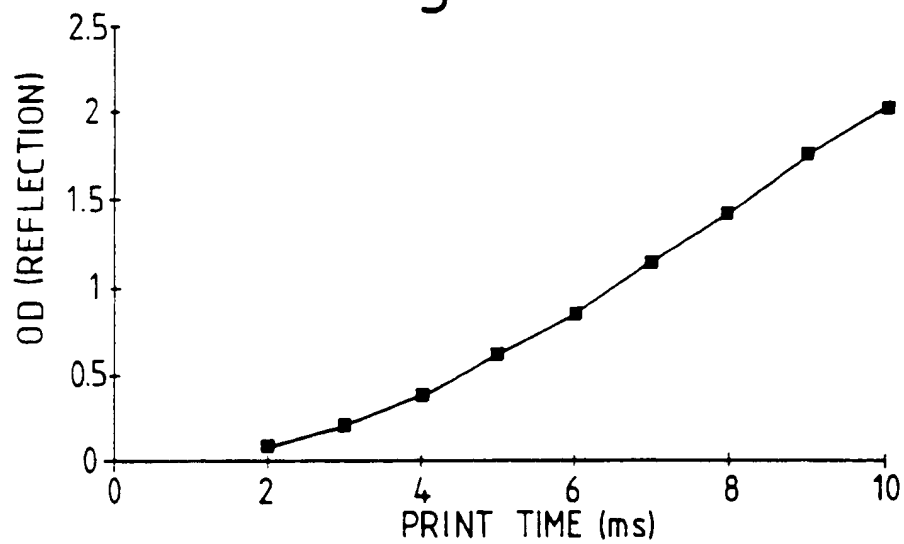
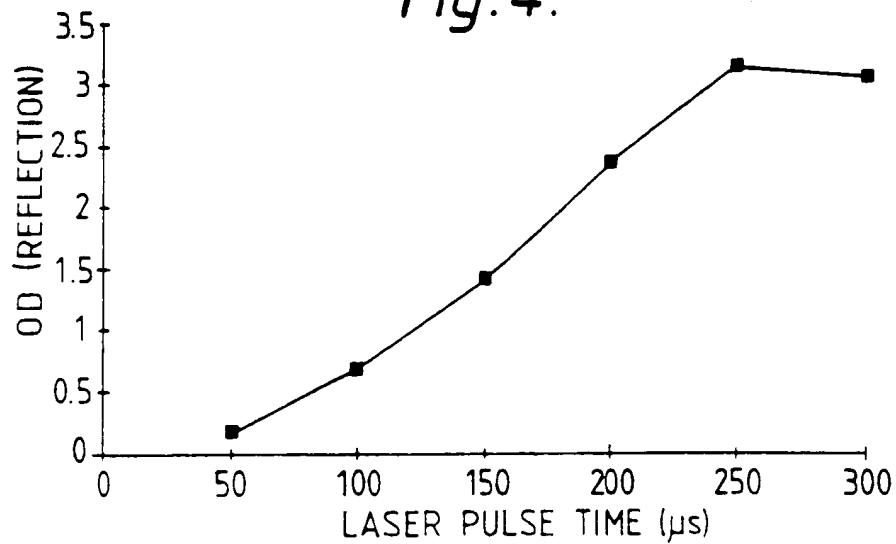
-

1/3

*Fig. 1.**Fig. 2.*

SUBSTITUTE SHEET

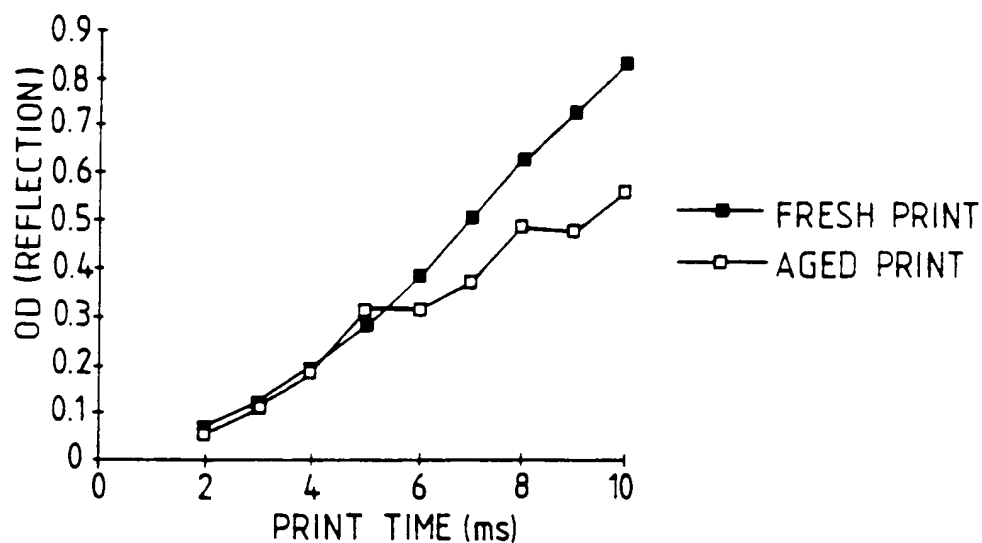
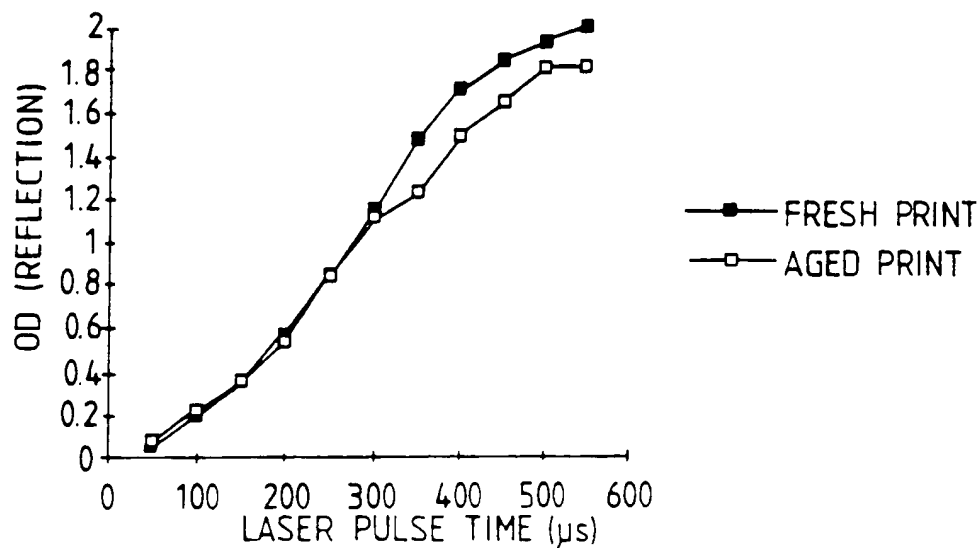
2/3

*Fig. 3.**Fig. 4.*

SUBSTITUTE SHEET



3/3

*Fig. 5.**Fig. 6.*

SUBSTITUTE SHEET

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>6</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl. 5 B41M5/40; B41M5/00; B41M5/38; G03F7/20		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>7</sup>		
Classification System	Classification Symbols	
Int.Cl. 5	B41M ; G03F	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>8</sup>		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT<sup>9</sup></b>		
Category <sup>10</sup>	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
X	EP,A,0 454 083 (EASTMAN KODAK COMPANY) 30 October 1991 see page 2, line 31 - line 42 see page 3, line 13 - line 15 see page 4, line 25 - line 29 ---	1-2
X	EP,A,0 365 222 (EASTMAN KODAK COMPANY) 25 April 1990 see page 3, line 40 - line 46; claim 1 ---	1-2
X	EP,A,0 409 598 (OJI PAPER COMPANY) 23 January 1991 see page 4, line 45 - line 52; claims 12,14 ---	2-5
X	GB,A,2 217 866 (OJI PAPER COMPANY) 1 November 1989 see page 8, line 25 - page 9, line 11 see page 11, line 16 - line 21 -----	2-5
<p><sup>10</sup> Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"C" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search		Date of Mailing of this International Search Report
13 AUGUST 1993		31. 08. 93
International Searching Authority		Signature of Authorized Officer
EUROPEAN PATENT OFFICE		BACON A.J. <i>A.J. Bacon</i>

**ANNEX TO THE INTERNATIONAL SEARCH REPORT  
ON INTERNATIONAL PATENT APPLICATION NO.**

GB 9301267  
SA 74987

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report.  
The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

13/08/93

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0454083	30-10-91	US-A- 5126760	30-06-92
		JP-A- 4226392	17-08-92
-----			
EP-A-0365222	25-04-90	JP-A- 2176708	09-07-90
		US-A- 4923860	08-05-90
-----			
EP-A-0409598	23-01-91	JP-A- 3049991	04-03-91
		JP-A- 3142285	18-06-91
		JP-A- 3142286	18-06-91
		JP-A- 3183591	09-08-91
		US-A- 5143904	01-09-92
-----			
GB-A-2217866	01-11-89	JP-A- 1263081	19-10-89
		JP-A- 1267090	24-10-89
		JP-A- 1271289	30-10-89
		US-A- 5177053	05-01-93
-----			

EPO FORM P0479

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

